



New Processes Development in Pharmaceutical R&D and Manufacture

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CARES 10-year anniversary 1/12/2023





Generic Challenges of Process Development in Pharma

- Synthesis of an active pharmaceutical ingredient (API) is typically a multi-step sequence of synthesis and purification tasks.
- There are different options of how to build the required molecule (different synthetic paths).
- There is a large number of combinations of possible reagents and reaction conditions.

An example of possible chemical routes from limonene (a biowaste based starting molecule) to paracetamol. In total there are 458 routes involving 132 chemical species. Each reaction (arrow) will have a set of optimal reaction conditions and many of the reactions must be followed by product separation (not shown).



Generic Challenges of Process Development in Pharma

- How to efficiently develop a single step (reaction or separation)?
- How to efficiently design the complete sequence of steps?



- Could we predict / calculate reaction outcome (what is being formed) and reaction conditions?
- Is it possible to generate required data using minimum time and quantities of reactants/reagents?
- Can we calculate or efficiently measure all required physical properties to evaluate separation strategy computationally?
- How to efficiently enumerate all possible options for a multi-step synthesis?
- What is the most effective way of optimizing a multi-step synthesis?

PIPS projects hosted by CARES

Pharma Innovation Programme Singapore

The Pharma Innovation Programme Singapore (PIPS) is an industry-led platform which aims to synergistically and strategically bring together public sector research capabilities and domain expertise of the pharmaceutical industry to enhance the productivity and operational efficiency within Singapore's pharmaceutical sector through leveraging novel manufacturing technologies and data analytics. CARES PIPS-1 Projects (completed in 2022/2023)

C4: Development of Multi-Step Processes in Pharma C12: Data-2-Knowledge in the Digital Manufacture of Pharmaceuticals Pfizer-specific:

CARES PIPS-2 Projects (started in 2023)

T1: From Digital Twins to Real Time Al-supported PlantOperationsT2: Automated Evaluation of Environmental Impacts ofPharma Manufacturing Processes

Efficient Development of a Single Reaction Step

Li-Halogen exchange is a frequently used synthetic strategy despite well-known practical difficulties:

- very high exotherm and reaction rate (requires cryogenic conditions);
- low solubilities solids formation;
- high sensitivity to traces of water and oxygen;
- low tolerance to functional groups formation of impurities.



Could we propose a robust procedure based on Li-halogen exchange and used at scale?

Dr Dogancan Karan

Efficient Development of a Single Reaction Step



Simplified reaction – focus on the most complex step of Li-halogen exchange.

Drs Dogancan Karan & Guoying Chen

Flow Chemistry for 'Difficult' Synthetic Procedures



Benefits of flow technology:

- Many options for mixing arrangement in flow.
- Excellent temperature control high exchange area.
- Enclosed system for control of environment.
- Low hazard small reaction volumes.
- High space-time-yield: no need for scale-up.



Drs Dogancan Karan & Guoying Chen

Engineering a Good Experiment

- CFD simulations to ensure that cooling tube is long enough to reach the desired temperature.
- Scaled down, axisymmetric 2D geometry to save computational time.
- o Highest flow speed (worst case scenario) is simulated.



3-3.2 m tube length is enough to reach target temperature



$$\begin{split} \rho(\boldsymbol{u}\cdot\nabla)\boldsymbol{u} &= \nabla\cdot[-p\boldsymbol{I}+\boldsymbol{K}]+\boldsymbol{F} \quad \rho\nabla\cdot\boldsymbol{u}=0 & \text{NSE for velocity field} \\ \rho C_p\boldsymbol{u}\cdot\nabla T+\nabla\cdot\boldsymbol{q} &= Q+Q_{ted} \quad \boldsymbol{q}=-\mathbf{k}\nabla T & \text{Heat transport at solid side} \\ \rho C_p\boldsymbol{u}\cdot\nabla T+\nabla\cdot\boldsymbol{q} &= Q+Q_p+Q_{vd} \quad \boldsymbol{q}=-\mathbf{k}\nabla T & \text{Heat transport at liquid side} \end{split}$$



Engineering a Good Experiment

- o Sequence of events to collect a data point causes dispersion in multiple locations in the system
- Steady state time needs to be estimated from the theory of Residence Time Distribution (RTD)
- o Dispersion regimes are identified from Ud_t/D_{AB} vs L/d_t plot





- Taylor dispersion model is used for n-BuLi+THF and sample collection part
- $\circ~$ Plug flow is assumed in reaction zone due to short τ
- Predicted steady state with dispersion model is 50 100 % higher than plug flow steady state time for 15 randomly generated samples
- o Fluctuation in steady state samples (avg of 10 samples)

Sample I - 47.3 (± 3.8) % Yield, 19.5 (± 1.1) % Impurity Sample II – 95.5 (± 0.5) % Yield, 4.7 (± 0.5) % Impurity

Reaction Development & Optimisation using Automated Experiments



An open-source solution





https://pypi.org/project/flab/

About

- A coding *framework* to simplify automation
 - Providing a robust architecture and tools for implementing routines
- Open-source *Python*
 - Zero capital cost, and native migration of many packages for ML & data-science
- Designed for experimentalists with novice coding experience
 - Providing a gradual transition into coding expertise
- Agile principles
 - Making modification and sharing a feature

Automation Workflow

- 1. Choose the devices you need, and load them into the automation framework
- 2. Program a task (or multiple tasks) in python with high-level commands
 - i.e. " device1.start()" or "experiment.start(parameters)"
- 3. Construct AI "loops"
- 4. Run and stop task(s) as you need during an experiment
- 5. Increase in complexity as necessary

Generating Process Knowledge through ML-driven DoE

Multi-objective Bayesian Optimization: Yield (max) vs Impurity (min) as competing objectives





Top 3 best conditions for max yield

τ (s)	T (°C)	n-BuLi (Eq)	Yield (%)	Imp (%)
0.191	-23.41	1.079	94.1	4.1
0.23	-30	1.047	93.3	3.4
0.185	-26	1.037	93.1	2.5

- TSEMO designed 34 experiments to establish the pareto plot
- o Optimum operation range for max. yield

ο
$$τ$$
 = 0.185-0.230 s, T = -(30-23) ºC, n-BuLi = 1.03 − 1.08 Eq.

Generating Process Knowledge through ML-driven DoE

Multi-objective Bayesian Optimization: Yield (max) vs Impurity (min) as competing objectives



- TSEMO designed 28 experiments to establish the pareto plot
- o Optimum operation range for max. yield
- ο τ = 0.185-0.266 s, T = -(30-23) °C, n-BuLi = 1.0 − 1.005 Eq.



Top 3 best conditions for ma	ax yield
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τ (s)	T (°C)	n-BuLi (Eq)	Yield (%)	Imp (%)
0.24	-28.6	1.003	95.7	3.6
0.266	-23.5	1.005	95.6	3.3
0.19	-23.4	1.06	93.9	5.57

Tubular reactor vs Microfluidic chip reactor



Tubular reactor vs Microfluidic chip reactor

Comparison between tubular system and chip system

Best performing condit	ion of different optim	ization campaigns
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Optimization	τ (s)	T (°C)	n-BuLi (Eq)	Yield (%)	Imp (%)
Tube+12 Training	0.191	-23.41	1.079	94.1	4.1
Tube+3 Training	0.185	-16.3	1.065	93.8	4.8
Chip+12 Training	0.24	-28.6	1.003	95.7	3.6

- Chip system can achieve higher yield and lower impurity due to better mixing
- o Tubular system requires more n-BuLi to achieve the max yield

	τ (s)	T (°C)	n-BuLi (Eq)	Yield (%)	Imp (%)
Tube	0.185	-21	1.1	91.9	8
Chip	0.19	-23.4	1.06	93.9	5.57

o Tubular system requires more n-BuLi to compromise the yield for impurity



Consistent Pareto efficient points for two optimization campaigns with tubular system Slightly better pareto efficient points for chip system due to improve mixing

Efficient data generation in an automated laboratory

C4 PIPS-1 Project: Development of Multi-Step Processes in Pharma

Project Objectives

- To develop a generic methodology for accelerated development of multi-step reaction-separation processes in the manufacture of pharmaceuticals.
- To demonstrate the methodology on specific industry-defined case studies.
- To identify critical technological challenges for implementation.

Industrial Case Study



Step 1



Entry	Continuous variables	Discrete variables
1	Reaction temperature, T_1	Type of BASE
2	Reaction time, t_{r1}	Type of SOLV 1
3	Equivalents of 6	
4	Equivalents of BASE	





Entry	Continuous variables	Discrete variables
1	Reaction temperature, T_2	Type of ACID
2	Reaction time, t_{r2}	Type of SOLV₂
3	Equivalents of 7	
4	Equivalents of ACID	

$SOLV_1$ Options

- Methanol
- Chloroform
- 1,3-Dimethyl-2-imidazolidinone
- Tetrahydrofuran
- N, N'-Dimethylpropyleneurea

BASE Options

- 1,8-Diazabicycloundec-7-ene
- 1,5-Diazabicyclo[4.3.0]non-5-ene
- N-Methylmorpholine
- N,N-Diisopropylethylamine
- Triethylamine
- 2,6-Lutidine
- Pyridine

ACID Options

- Acetic Acid
- Cyclobutanecarboxylic acid
- Cyclopentanecarboxylic acid
- Difluoroacetic acid
- Formic acid
- Isobutyric acid
- Isovaleric acid
- Lactic acid
- Pentanoic acid
- Propionic acid

Machine-learning based design of experiments: TSEMO

Optimisation of STY (g/L/h) and E-Factor (Reaction Mass/Product Mass)



Drs Mohammed Jeraal, Simon Sung, Magda Barecka

- Optimisation space same as original
- Same training experiments as original optimisation
- System performed 74 optimisation experiments.

M.I. Jeraal, S. Sung, A.A. Lapkin, A Machine Learning-Enabled Autonomous Flow Chemistry Platform for Process Optimization of Multiple Reaction Metrics. Chemistry– Methods, 2 (2021) 71-77

User interface for working with automated experiments

Single-Step Multi-Objective Optimisation Station

Step 1:	Setup Interface with Flow Chemis	try Equipment	Step 4:	Select Reaction Scale	Training	Optimisation	
Launch Flow Commander		Select the limiting quantity fro	m Pump A for each run (ml) 5	Exp No:	Exp. No:	n/a	
k blue arrow icon t	to ensure all components are "online" b	efore connecting.					
Connect	Connected to local instance of Flo	w Commander	Step 5:	Select Optimisation Targets	Experimental Conditions:	Current Last F	Reacti
			Select the target functions for	the multi-objective optimisation from the list below.			
Step 2:	Define Optimisation Sp	ace	Target Objective 1: [Select]	▼ Target Objective 2: [Select] ▼			
ect and edit optimis	stion targets. [Please delete any unnec	essary variables].	If prompted, enter the relevan	t chemical properties upon confirmation.			
	Select Optimisation Variables		Confirm Objective Func	tions			
	Lower Limit	Upper Limit					
	0	0	Step 6:	Train Optimisation Model			
	0	0	Manual Enter ex	sting training data into Excel file and save below.	Reaction Outputs:	Optimal Last F	React
	0	0	Generate Excel File	Save External Training Data			
	0	0	Auto Enter no	of experiments per variable: 8			
	0	0	Generate Auto Experin	nents			
Confirm L	Limits			Start Training			
Step 3:	Specify Chemical Inp	ıts			Generate Graph of	Optimisation Experiments	
cify the required p	properties of chemical species in the sys	tem.	Step 7:	Machine Learning Optimisation			
Reactant So Analytical Pa	olutions		s	tart Optimisation	Pause Resume	System Running	g
Status: Setup Optimisation Setup. Please proceed through the steps to comm			ease proceed through the steps to	commence optimisation.	Last Status:	10-Feb-2020 10:38:11	

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A chemical robot = an automated Vapourtec flow reactor + Liquid Handler + HPLC

Components:

- Flow chemistry experiment precise control of conditions, broad range and rapid change of operating conditions
- Reactions in slugs to conserve material and increase data throughput
- Generation of reaction mixtures by input liquid handling robot



<u>N</u>omadic <u>E</u>volutionary <u>M</u>ultiobjective <u>O</u>ptimisation (NEMO)



to refine the locations

Mohammed Jeraal, Simon Sung

Advance in algorithmic optimisation of reactions



NEMO improves search for optimal conditions in multi-objective optimisation

Prediction of reaction impurities in chemical reactions



Adarsh Arun

Prediction of reaction impurities in chemical reactions







A. Arun, Z. Guo, S. Sung, A.A. Lapkin, Reaction Impurity Prediction using a Data Mining Approach. Chemistry Methods, 3 (2023) e2022000062.

Ongoing PIPS projects hosted by CARES

T1: From Digital Twins to Real Time AI-supported Plant Operation

CARES:

Alexei Lapkin (PI) Markus Kraft

I2R A*STAR: Lianlian Jiang



Ongoing PIPS projects hosted by CARES

T2: Automated Evaluation of Environmental Impacts of Pharma Manufacturing Processes



constraints and estimates based on mechanistic models: more data

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Spin-out company offering consultancy and developing a new product (cdi-sg.com)



Centre for Doctoral Training @ University of Cambridge (syntechcdt.com)



Innovation Centre to work with SMEs hosted by the University of Cambridge (idmt.online)



Early career researchers

Dr Zhen Guo Dr Simon Sung Dr Mohammed Jeraal Dr Dogancan Karan Dr Buoying Chen Dr Magda Barecka Dr Shuyuan Zhang Dr Jiyizhe Zhang Kobi Felton Adarsh Arun Jiaru Bai (MK)